### Application Note · ZEEnit 700P



#### Challenge

Reliable and cost-effective heavy metal analysis in matrixrich sewage sludges and soils.

#### Solution

Automated elemental analysis with the ZEEnit 700P combining flame and furnace technique for ideal suitability in the required concentration range.

#### Intended audience

Private and state environmental laboratories as well as wastewater treatment plants.

# Standardized Heavy Metal Determination in Sewage Sludges and Soils Using Atomic Absorption Spectrometry (AAS)

### Introduction

Sewage sludge is generated as a waste product of wastewater treatment and consists of organic and mineral substances in enriched form. Thus, especially plant nutrients, such as phosphate and nitrate are present in higher concentrations. Sewage sludge can be used in agriculture as a primary marterial for the fertilizer production. One of the concerns is the content of heavy metals in sewage sludge. Of concern here are heavy metals in sewage sludges, which are released into the environment, where they eventually migrate into plants and conterminating the ground water. Hence, apart from monitoring heavy metal contents in sewage sludge, their determination in soils is also a routine task.

The reuse of sewage sludge is regulated by national directives and ordinances, in Germany by the sewage sludge ordinance (AbfKlärV). The aim of the regulations is the reduction of the anthropogenic emission of contaminants into the environment to a neglectable level. For the elements lead, cadmium, chromium, copper, nickel, and zinc (sometimes also manganese and cobalt) the analysis

by Atomic Absorption Spectrometry (AAS) is specified in multiple norms (e.g., DIN ISO 11047 or DIN EN ISO 15586). AAS is a simple, robust, cost effective, and well established technique used among others by wastewater treatment plants, in governmental institutions, or contract laboratories. This application note describes the analysis of sewage sludge by flame as well as graphite furnace atomic absorption spectrometry. Using the AAS ZEEnit 700P, both atomization techniques can be performed without modification. Advantages are offered by the ZEEnit series due to its Zeeman background correction with a variable magnetic field strength up to one Tesla, which can be operated in 2-field, 3-field and dynamic mode. The dynamic mode in graphite furnace technique combines the 2-field and the less sensitive 3-field mode and, therefore, automatically enables an enlarged measuring range. Samples with very high matrix loads such as sewage sludge can thus be easily analyzed with excellent background correction and without additional dilution steps.



### Materials and Methods

Heavy metal analysis by ZEEnit 700P is shown on the example of certified reference material. All operating steps as well as the purity of the reagents and used materials correspond to the guidelines of the (ultra) trace analysis.

#### Samples and reagents

- Concentrated HNO<sub>3</sub> (65%)
- Concentrated HCI (37%)
- Cesium chloride-lanthanum chloride (Cs/La) buffer solution according to Schinkel (10 g/L CsCl, 100 g/L LaCl<sub>3</sub>)
- Mg-modifier (10 g/L Mg(NO<sub>3</sub>)<sub>2</sub>)
- Pd-modifier (10 g/L Pd)
- Ascorbic acid
- Certified single element standards for Cr, Mn, Co, Ni, Cu, Zn, Cd, and Pb (Concentration of the analyte respectively 1000 mg/L)
- Reference material:
  - BCR-143R sewage sludge with enriched soil (European Commission, Institute for Reference Materials and Measurements)
  - BCR-146R industrial sewage sludge (European Commission, Institute for Reference Materials and Measurements)
  - BAM-U110 contaminated soil (BAM, German federal institute of material research and testing, 2006)

#### Sample preparation

The samples were digested according to the DIN ISO 11466 using aqua regia. The weight of the sample material was approx. 0.3 g. The digesting solution was filled up to 50 mL with the corresponding diluent.

Sample preparation for flame measurement is based on the DIN ISO 11407. For the dilution of standards and samples

even lower acid concentrations than listed in the norm stabilize the measurement solutions adequately. For measurements by flame technique the samples were diluted with a solution containing 21 % (v/v) concentrated HCl and 7 % (v/v) concentrated HNO<sub>3</sub>. For the elements chromium and manganese, an additional 10 % (v/v) Cs/La buffer solution was added.

The sample dilution for the measurement by graphite furnace technique is based on the DIN EN ISO 15586. The corresponding diluent was 1 % (v/v) HNO<sub>3</sub>.

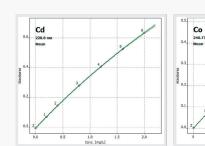
#### Calibration for flame technique

In accordance with the DIN ISO 11047, the calibration standards were prepared in a solution containing 21 % (v/v) HCl and 7 % (v/v) HNO<sub>3</sub>. Such a solution with 21 % (v/v) HCl and 7 % (v/v) HNO<sub>3</sub> was also used as the blank value for the calibration. For the elements chromium and manganese, an additional 10 % (v/v) Cs/La buffer solution was added. All standards were prepared from certified single element standards at a concentration of 1000 mg/L. The calibration functions for the flame technique are shown in Figures 1 to 8. The concentrations of the applied calibration standards are given in table 1.

According to the DIN ISO 11047, lead is measured at the wavelength 217 nm. Alternatively, the wavelength 283 nm is recommended. The slightly less sensitive absorption line at 283 nm provides a better signal-to-noise ratio and reduced unspecific absorption.

Standard	Cd [mg/L]	Co [mg/L]	Cr [mg/L]	Cu [mg/L]	Mn [mg/L]	Ni [mg/L]	Pb [mg/L]	Zn [mg/L]
CalStd. 0	0	0	0	0	0	0	0	0
CalStd. 1	0.2	1	1	1	0.4	1	1	0.2
CalStd. 2	0.4	2	2	2	1	2	2	0.4
CalStd. 3	0.8	4	4	4	2	4	4	0.8
CalStd. 4	1.2	6	6	6	4	6	6	1.2
CalStd. 5	1.6	8	8	8	6	8	8	1.6
CalStd. 6	2.0				8			2.0

Table 1: Concentration of the calibration standard for flame technique





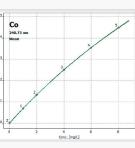


Figure 2: Co  $R^{2}_{(adj)} = 0.9999$  (non-linear)

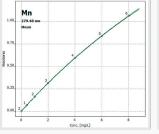


Figure 5: Mn (Acetylene/Nitrous oxide)  $R^{2}_{(adj)} = 0.9999$  (non-linear)

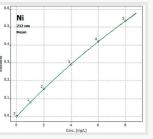


Figure 6: Ni  $R^{2}_{(adj)} = 1.000$  (non-linear)

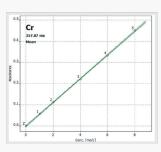


Figure 3: Cr (Acetylene/Nitrous oxide)  $R^{2}_{(adj)} = 0.9998$  (linear)

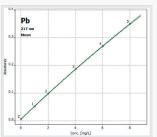


Figure 7a: Pb (217 nm)  $R^2_{(adj)} = 1.0000$  (non-linear)

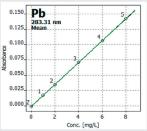




Figure 7b: Pb (283 nm)  $R^{2}_{(adj)} = 1.0000$  (linear)

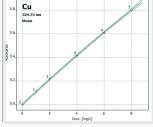


Figure 4: Cu  $R^{2}_{(adj)} = 0.9998$  (non-linear)

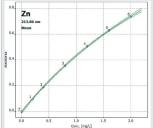


Figure 8: Zn  $R^{2}_{(adj)} = 0.9997$  (non-linear)

#### Calibration for graphite furnace technique

In accordance with DIN EN ISO 15586, the calibration standards were prepared as shown in table 2. The calibration standards were stabilized with diluted nitric acid (1 % (v/v) concentrated  $HNO_3$ ). The diluted  $HNO_3$  solution (1 % (v/v) HNO<sub>2</sub>) was used as blank solution and as diluent. A Pd-Mg solution  $(1 \text{ g/L Pd}, 0.5 \text{ g/L Mg}(NO_3)_2)$ , a Mg solution (0.5 g/L Mg(NO<sub>3</sub>)<sub>2</sub>), and a solution containing  $(NH_4)H_2PO_4$  (1 % (w/w)) were used as matrix modifiers.

For better stabilization of highly volatile chloride compounds of the analytes, ascorbic acid can be used in addition to the Pd/Mg solution to pre-reduce the palladium. This can be automatically injected into the graphite tube by the autosampler. As a simultaneous mixing in a sample vessel will cause the palladium to precipitate. Therefore, the autosampler is capable of a contact-free uptake of these solutions in one aspiration step. This can be programmed in the software as an option in the method parameters.

Standard	Cd [µg/L]	Co [µg/L]	Cr [µg/L]	Cu [µg/L]	Mn [µg/L]	Ni [µg/L]	Рb [µg/L]
CalStd. 0	0	0	0	0	0	0	0
CalStd. 1	0.4	6	2	3	1.5	7	1
CalStd. 2	1.2	18	6	9	4.5	21	2
CalStd. 3	2.0	30	10	15	7.5	35	4
CalStd. 4	2.8	42	14	21	10.5	49	6
CalStd. 5	4.0	60	20	30	15.0	70	8

Table 2: Concentration of the calibration standards for graphite furnace technique

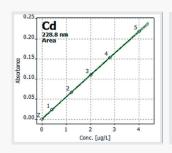


Figure 9: Cd  $R^{2}_{(adj)}$  = 0.9999 (linear)

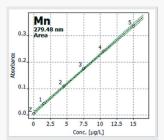


Figure 13: Mn R<sup>2</sup><sub>(adj)</sub>= 0.9991 (linear)

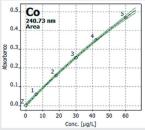


Figure 10: Co  $R^{2}_{(adj)}$ = 0.9995 (non-linear)

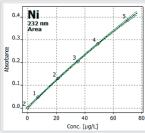


Figure 14: Ni  $R^{2}_{(adj)}$ = 0.9997 (non-linear)

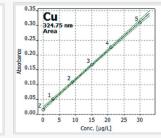


Figure 11: Cu  $R^2_{(adj)} = 0.9990$  (linear)

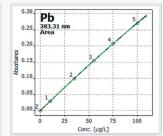


Figure 15: Pb (283 nm) R<sup>2</sup><sub>(adj)</sub>= 0.9999 (non-linear)

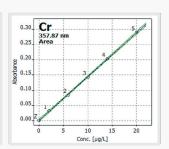


Figure 12: Cr  $R^{2}_{(adj)} = 0.9995$  (linear)

#### Instrumentation

The analysis was performed using both flame technique and graphite furnace technique of the ZEEnit 700P. The instrument parameters are summarized in table 3 to 6. Element-specific temperature programs are used for the graphite furnace technique. These are shown in table 7. Table 3: General instrument specification for flame technique

Parameter	Specification
Device	ZEEnit 700P
Burner type and position	50 mm, 0°
Flame type	Acetylene/air; acetylene/nitrous oxide
Rinsing solution	2 % (v/v) HCl

Element	Wavelength [nm]	Slit [nm]	Lamp current [mA]	Flame type	Fuel gas flow [L/h]	Burner height [mm]	Background correction
Cd	228.8	1.2	2	Air/C <sub>2</sub> H <sub>2</sub>	45	4	D <sub>2</sub> -HCL
Со	240.7	0.2	6	Air/C <sub>2</sub> H <sub>2</sub>	65	6	D <sub>2</sub> -HCL
Cr	357.9	0.2	5	Air/ $C_2H_2$ N <sub>2</sub> O/ $C_2H_2$	95 185	10 4	-
Cu	324.8	1.2	2	Air/C <sub>2</sub> H <sub>2</sub>	45	5	D <sub>2</sub> -HCL
Mn	279.5	0.2	б	Air/C <sub>2</sub> H <sub>2</sub> N <sub>2</sub> O/C <sub>2</sub> H <sub>2</sub>	80 180	6 4	D <sub>2</sub> -HCL
Ni	232.0	0.2	5	Air/C <sub>2</sub> H <sub>2</sub>	45	5	D <sub>2</sub> -HCL
Pb	217.0 283.3	1.2	4	Air/C <sub>2</sub> H <sub>2</sub>	60	6	D <sub>2</sub> -HCL
Zn	213.9	0.5	2	Air/C <sub>2</sub> H <sub>2</sub>	45	4	D <sub>2</sub> -HCL

Table 4: Method settings for flame technique

Table 5: General instrument specification for graphite furnace technique

Parameter	Specification
Device	ZEEnit 700P
Tube type	PIN platform
Modifier	Pd/Mg (1 g/L Pd, 0,5 g/L Mg $(NO_3)_2$ ) 1 % (w/w) ascorbic acid Mg (0.5 g/L Mg $(NO_3)_2$ ) NH <sub>4</sub> H <sub>2</sub> PO <sub>4</sub> (1 % (w/w))
Repeat measurements	3
Evaluation	Area
Background correction	Zeeman (1 T)
Rinsing solution	$1 \% (v/v) HNO_3$ 0.05 % (w/w) tergitol

Element	Wavelength [nm]	Slit [nm]	Lamp current [mA]	Modifier	Magnetic field strength [T]
Cd	228.8	0.8	2	$NH_4H_2PO_4$	1
Со	240.7	0.2	5	Mg	1
Cr	357.9	0.2	4	Mg	0.7
Cu	324.8	0.8	2	Pd/Mg*	1
Mn	279.5	0.2	5	Pd/Mg*	1
Ni	232.0	0.2	4	Mg	1
Pb	283.3	0.8	3	NH <sub>4</sub> H <sub>2</sub> PO <sub>4</sub>	1

### Table 6: Method settings for graphite furnace technique

\* addition of ascorbic acid for Pd reduction

Table 7: Furnace program for graphite furnace technique

Element	Temp. [°C]	Ramp [°C/s]	Hold [s]	Element	Temp. [°C]	Ramp [°C/s]	Hold [s]
Cd	85	6	15	Co	85	6	15
	95	3	15		95	3	15
	110	5	10		110	5	10
	350	50	5		350	50	5
	550	300	10		950	300	10
	1450	1400	3		2200	1500	4
	2450	500	4		2450	500	4

Element	Temp. [°C]	Ramp [°C/s]	Hold [s]	Element	Temp. [°C]	Ramp [°C/s]	Hold [s]
Cu	85	6	15	Mn	85	6	15
	95	3	15		95	3	15
	110	5	10		110	5	10
	350	50	5		350	50	5
	950	300	10		700	300	10
	2100	1500	4		1600	1500	4
	2450	500	4		2450	500	4

Temp. [°C]	Ramp [°C/s]	Hold [s]	Element	Temp. [°C]	Ramp [°C/s]	Hold [s]
85	6	15	Pb	85	6	15
95	3	15		95	3	15
110	5	10		110	5	10
350	50	5		350	50	5
1000	300	10		700	300	10
2450	1500	5		1600	1400	3
2550	500	4		2450	500	4
	[°C] 85 95 110 350 1000 2450	[°C/s]   85 6   95 3   110 5   350 50   1000 300   2450 1500	[°C/s][s]8561595315110510350505100030010245015005	[°C/s][s]8561595315110510350505100030010245015005	[°C]     [°C]     [°C]       85     6     15     Pb     85       95     3     15     95     10     110       350     50     5     350     350     350     350       1000     300     10     700     1600     1600     1600	I'C)I'C/sI'C/s $85$ 615Pb $85$ 6 $95$ 315 $95$ 3 $110$ 510 $110$ 5 $350$ 505 $350$ 50 $100$ $300$ 10 $700$ $300$ $2450$ $1500$ 5 $1600$ $1400$

### Results and Discussion

The results of heavy metal determination in the certified reference materials are shown in table 8. No or only low dilution factors were applied using the flame technique. Using the more sensitive graphite furnace technique a stronger dilution of the samples was necessary for most elements.

The quantification of chromium and manganese may show interferences with the air-acetylene flame. This can be compensated by an increased addition of Cs/La solution or by adding of NaSO<sub>4</sub> (e.g., 0.1 % (w/w)). Alternatively, analysis can be performed using a nitrous oxide-acetylene flame. With this type of flame the occurring interferences of the air-acetylene flame are hardly or not at all recognizable. The relevant concentrations of zinc in environmental samples clearly exceed the measuring range of the graphite furnace technique. Therefore, this element is preferably

quantified by means of the flame technique. With the ZEEnit series, the 3-field mode can be used to reduce the measurement sensitivity for zinc to a level where environmentally relevant issues can also be measured in the graphite furnace technique without using high dilution factors, high purified reagents, and acids. After aqua regia digestion, the cobalt values of the reference material BCR 143R and BCR 146R were already close to the determination limit of the flame technique. Therefore, the graphite furnace technique was carried out here. For the determination of lead in sewage sludge and soil, the absorption line at 283 nm was used. At this wavelength, background absorption is lower, which is advantageous for samples with matrix loads. In addition, better detection limits are achieved at 283 nm.

Sample: BCF	R 143R	R Flame mode		Graphite furnace	Graphite furnace mode			
Element	Cert. value [mg/kg]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]	
Cr*	426 ± 12	1	437 ± 7.7	103	-	-	-	
Mn*	858 ± 11	2	836.3 ± 3.4	97	500	872 ± 12	102	
Со	12.3 ± 0.3	-	-	-	2	11.33 ± 1.2	92	
Ni	296 ± 4	1	302.1 ± 1.1	102	100	291.4 ± 5.4	98	
Cu	128 ± 7	1	131.9 ± 0.7	103	100	128.6 ± 12	100	
Zn	$1063 \pm 16$	10	1042 ± 2.4	98	-	-	-	
Cd	72 ± 1.8	2	70.1 ± 0.7	97	250	70.33 ± 2.8	99	
Pb	174 ± 4	1	168.2 ± 1.6	97	50	176.5 ± 1.4	101	

Table 8: Results for soil and sewage sludge reference material

Sample: BCR	Sample: BCR 146R Flame mode		Graphite furnace mode				
Element	Cert. value [mg/kg]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]
Cr	174 ± 4	1	172.7 ± 0.1	99	100	172.6 ± 1.6	99
Mn	298 ± 9	2	285.6 ± 3.1	96	300	307.5 ± 2.5	103
Со	6.5 ± 0.4	-	-	-	2	6.20 ± 0.07	95
Ni	65 ± 3	1	61.2 ± 0.8	94	10	62.1 ± 1.0	96
Cu	831 ± 16	1	819.2 ± 1.9	99	100	821 ± 2.3	101
Zn	3040 ± 60	30	3020 ± 11	99	-	-	-
Cd	18.4 ± 0.4	2	17.91 ± 0.16	97	100	17.92 ± 0.23	97
Pb	583 ± 17	1	560.5 ± 6.4	96	100	588.7 ± 6.4	101

Sample: BAN	A U110	Flame mode			Graphite furnace mode		
Element	Cert. value [mg/kg]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]	Dilution factor	Meas. value [mg/kg]	Recovery rate [%]
Cr	190 ± 9	1	$184.4 \pm 0.7$	97	-	-	-
Mn	580 ± 19	2	546.7 ± 1.2	94	400	586.2 ± 3.2	101
Со	14.5 ± 0.8	1	13.4 ± 1.1	92	3	14.27 ± 0.11	98
Ni	95.6 ± 4	1	91.7 ± 1.4	96	20	92.8 ± 0.8	97
Cu	262 ± 9	1	262.1 ± 2.5	100	100	252 ± 2.7	96
Zn	990 ± 40	10	948.2 ± 4.5	96	-	-	-
Cd	7 ± 0.4	1	6.65 ± 0.015	95	50	6.93 ± 0.24	99
Pb	185 ± 8	1	191.7 ± 3.1	103	50	185.2 ± 3.2	100

\*analyzed with acetylene-nitrous oxide flame

 $Table \ 9: Detection \ limits \ for \ flame \ technology \ determined \ according \ to \ the \ 3\sigma-criterion \ of \ an \ 11-fold \ blank \ value \ measurement$ 

Element	Wavelength [nm]	Detection limit [mg/L]
Cd	228.8	0.0040
Со	240.7	0.0070
Cr	357.9	0.014 (air-acetylene)
		0.0095 (nitrous oxide-acetylene)
Cu	324.8	0.0030
Mn	279.5	0.0040 (air-acetylene)
		0.0123 (nitrous oxide-acetylene)
Ni	232.0	0.0063
Pb	217.0	0.062
	283.3	0.010

Element	Wavelength [nm]	Detection limit [µg/L]
Cd	228.8	0.0090
Со	240.7	0.24
Cr	357.9	0.060
Cu	324.8	0.11
Mn	279.5	0.044
Ni	232.0	0.41
Pb	283.3	0.30

Table 10: Detection limits for graphite furnace technology determined according to the 3 $\sigma$ -criterion of an 11-fold blank value measurement

## Summary

Thanks to its double sample chamber concept the ZEEnit 700P offers the usage of flame technique as well as graphite furnace technique without conversion. As the listed measurements show, the less sensitive flame technique is already sufficient enough for the determination of heavy metal concentrations in sewage sludge and contaminated soil samples. For lower concentrations of heavy metals, the graphite furnace technique with its significantly higher detection strength is required for precise measurement results. The exemplary selected elements lead, cadmium, chromium, copper, nickel, manganese, cobalt, and zinc could be determined very reliably with both techniques in the environmentally relevant reference materials showing excellent recovery rates of 92-103%. Thanks to the powerful Zeeman background correction of the ZEEnit 700P, even very low levels of toxic elements such as lead and cadmium can be easily and reliably quantified even in samples with high matrix loads.

To simplify analysis procedures in routine operations, autosamplers autosamplers with an automatic standard preparation and sample dilution function are available for both flame technique and graphite furnace technique to automate the analysis procedure.



Figure 16: ZEEnit 700P with autosampler

#### Overview of devices and consumables

Table 11: Required devices, accessories, and consumables

Article	Article number	Description
ZEEnit 700P	813-0700P-2-K	Flame- and graphite furnace AAS with Zeeman background correction with furnace vision tool
AS-GF Autosampler	-	Autosampler with auto-dilution for graphite furnace, included in the system
Chiller	810-60053-0	Software-controlled cooling system for graphite furnace technique, 50 Hz (60 Hz on request)
Consumable Set G	810-60079-0	Consumables for graphite furnace ZEEnit
AS-FD	810-60501-0	Autosampler with auto-dilution for flame
Air compressor	810-60055-0	To supply AAS with water- and oil-free compressed air, 50 Hz (60Hz on request)
50 mm Burner head	810-60057-0	Burner head for acetylene/nitrous-oxide or acetylene/air flame (100 mm burner head on request)
Scraper	810-60127-0	Automatic cleaning tool for 50 mm burner head
Consumable set F	810-60258-0	Consumables for flame of novAA/contrAA/ZEEnit-series
HCL Cd	480-450.008C	Coded hollow cathode lamp Cadmium
HCL Co	480-450.013C	Coded hollow cathode lamp Cobalt
HCL Cr	480-450.012C	Coded hollow cathode lamp Chromium
HCL Cu	480-450.014C	Coded hollow cathode lamp Copper
HCL Mn	480-450.032C	Coded hollow cathode lamp Manganese
HCL Ni	480-450.036C	Coded hollow cathode lamp Nickel
HCL Pb	480-450.028C	Coded hollow cathode lamp Lead
HCL Zn	480-450.067C	Coded hollow cathode lamp Zinc

Further HCL on request

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